

Acerosterol, a Novel Polyhydroxylated Sterol from the Gorgonian Octocoral *Pseudopterogorgia acerosa*

Lisa M. D. John, Winston F. Tinto, Stewart McLean, and William F. Reynolds

J. Nat. Prod., **1993**, 56 (1), 144-146 • DOI:
10.1021/np50091a023 • Publication Date (Web): 01 July 2004

Downloaded from <http://pubs.acs.org> on April 4, 2009

More About This Article

The permalink <http://dx.doi.org/10.1021/np50091a023> provides access to:

- Links to articles and content related to this article
- Copyright permission to reproduce figures and/or text from this article



ACS Publications
High quality. High impact.

Journal of Natural Products is published by the American Chemical Society, 1155 Sixteenth Street N.W., Washington, DC 20036

ACEROSTEROL, A NOVEL POLYHYDROXYLATED STEROL FROM THE GORGONIAN OCTOCORAL *PSEUDOPTEROGORGIA ACEROSA*LISA M.D. JOHN, WINSTON F. TINTO,*¹

Centre for Natural Products Chemistry, University of Guyana, Georgetown, Guyana

STEWART MCLEAN, and WILLIAM F. REYNOLDS*

Department of Chemistry, University of Toronto, Toronto M5S 1A1, Canada

ABSTRACT.—A new tetrahydroxysterol, acerosterol [2], has been isolated from the gorgonian octocoral *Pseudopterogorgia acerosa*. The structure has been determined to be 4 α ,23,24(R)-trimethyl-5 α -cholest-22E-ene-1 α ,3 β ,6 β ,11 α -tetraol on the basis of 2D nmr spectroscopy.

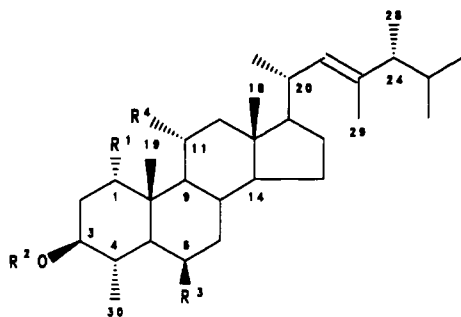
Gorgonian corals of the genus *Pseudopterogorgia* are a rich source of biologically active compounds (1). We previously reported the isolation and structure elucidation of some novel pseudopterane diterpenoids from the Caribbean sea whip *Pseudopterogorgia acerosa* Pallas (Gorgonidae) (2–5). We have further investigated the extracts of this organism and report here the isolation and structure determination of a new tetrahydroxysterol, designated acerosterol [2], from the most polar fractions.

Acerosterol [2], C₃₀H₅₂O₄, was isolated as white needles, and had an ir absorption at 3440 cm⁻¹ (OH). The ¹H-nmr spectrum had resonances due to one

olefinic, two tertiary, and five secondary methyl groups. The presence of a trisubstituted double bond was revealed by a one-proton resonance at δ 4.87 (dd, J = 1.1 and 9.3 Hz) and ¹³C signals at δ 131.32 (d) and 135.72 (s).

The ¹³C-nmr spectrum displayed signals due to four oxymethine carbons at δ 74.69, 71.48, 67.35, and 66.79, while a ¹³C-¹H shift correlated experiment showed the corresponding ¹H-nmr resonances at δ 3.92 (dd, J = 3.0 and 3.2 Hz), 3.45 (ddd, J = 5.0, 10.0, and 11.5 Hz), 3.95 (ddd, J = 5.0, 10.0, and 10.0 Hz), and 4.20 (ddd, J = 2.8, 3.0, and 5.7 Hz). On acetylation, 2 afforded the tetraacetate 3. These results suggested that acerosterol was a tetrahydroxy sterol, probably a dinostane derivative (6).

A ¹H-¹H 2D correlated spectrum revealed that the oxymethine protons at δ 3.92, 3.45, and 4.20 were attached to C-1, C-3, and C-6 of the sterol skeleton. These observations were confirmed by the use of a long-range ¹³C-¹H shift correlated experiment using our FLOCK pulse sequence (7). In this regard, C-3 (δ 71.48) showed three-bond correlations with H-1 (δ 3.92) and H-30 (δ 1.04), while C-7 (δ 39.98) and C-8 (δ 29.01) showed correlations with H-6 (δ 4.20). The fourth hydroxyl group (δ 3.95) was placed at C-11 (δ 67.35) based on the observation that this carbon had long-range correlations with H-9 and H-12. The assignments for C-9 and C-12 are



- 1 R¹=R²=R³=R⁴=H
- 2 R¹=R³=R⁴=OH, R²=H
- 3 R¹=R³=R⁴=OAc, R²=Ac

¹Present address: Department of Chemistry, University of the West Indies, Cave Hill Campus, Barbados.

secure since they showed three-bond correlations to H-19 and H-18, respectively. The stereochemistry at the relevant positions in the nucleus followed from the coupling constants. The structure of the side chain was also revealed from the COSY and FLOCK experiments. The stereochemistry at C-20 and C-24 in **2** is assumed to be the same as in dinosterol [**1**] (6,8,9) and related compounds (10). These results are summarized in Table 1 and led to the structure **2** for acerosterol.

The tetraacetate **3**, [α]_D -22.0°, had the molecular formula C₃₈H₆₀O₈ on the basis of hreims. The assignment of the carbons and protons (Table 1) followed from the HETCOR and FLOCK experiments.

Dinosterol [**1**] and other derivatives bearing a 4 α -methyl group are characteristic sterols of dinoflagellates (10). They have also been obtained from marine invertebrates containing symbiotic zooxanthellae (11). It seems likely therefore, that acerosterol [**2**] might have originated from symbiotic zooxanthellae present in *P. acerosa*.

EXPERIMENTAL

GENERAL EXPERIMENTAL PROCEDURES.—Mp's were taken on a Kofler hot stage apparatus and are uncorrected. Ir spectra were obtained on a Nicolet 5DX FTIR spectrometer using KBr discs. Optical rotations were measured on a Perkin-Elmer 243B polarimeter. The nmr spectra were recorded on a Varian XL-400 spectrometer operating at 400 MHz for protons and 100.6 MHz for carbons, with TMS as the internal stan-

TABLE 1. ¹³C and ¹H Assignments for Acerosterol [**2**]^a and Tetraacetate **3**.^b

Position	Compound 2		Observed 2- or 3-bond Connectivity	Compound 3 ^c	
	δ_C	δ_H		δ_C	δ_H
1	74.69	3.92(3.0, 3.2)	1.14, 1.95	76.81	4.74(2.9, 3.2)
2	37.05	1.95, 1.76	3.92	30.42	2.23, 1.63
3	71.48	3.45(5.0, 10.0, 11.5)	3.92, 1.95, 1.76, 1.04	73.67	4.62(5.3, 11.5, 11.5)
4	36.18	1.68	1.95, 1.34, 1.04	33.37	1.88
5	46.68	1.34	3.92, 1.83, 1.14, 1.04	45.28	1.61
6	66.79	4.20(2.8, 3.0, 5.7)	1.83	69.24	5.25(3.0, 3.0, 5.5)
7	39.98	1.83, 1.12	4.20	35.38	1.92, 1.17
8	29.01	1.77	4.20, 1.19	29.06	2.00
9	53.27	1.19	2.28, 1.83, 1.14	49.10	1.64
10	41.39	—	4.20, 1.34, 1.14	40.20	—
11	67.35	3.95(5.0, 10.0, 10.0)	2.28, 1.19	71.46	5.20(4.6, 9.3, 10.7)
12	50.70	2.28, 1.23	0.69	46.55	2.18, 1.17
13	42.86	—	0.69	42.60	—
14	54.95	1.11	0.69	55.22	1.18
15	24.35	1.56, 1.05		24.09	1.54, 1.08
16	28.07	1.72, 1.16	1.20	27.85	1.70, 1.18
17	56.76	1.20	0.94, 0.69	56.68	1.19
18	13.04	0.69	1.20	13.28	0.80
19	16.62	1.14	1.34, 1.19	16.31	1.18
20	34.56	2.33	0.94	34.38	2.31
21	20.60	0.94(6.1)		20.56	0.88(6.5)
22	131.32	4.87(1.1, 9.3)	1.49	131.02	4.83(1.2, 8.6)
23	135.72	—	1.49, 0.92	135.91	—
24	50.24	1.64	4.87, 1.49, 0.92, 0.76	50.20	1.63
25	30.80	1.51	0.82, 0.76	30.73	1.51
26	21.75	0.76(6.6)	0.82	22.03	0.77(6.6)
27	20.11	0.82(6.5)	0.76	20.11	0.83(6.4)
28	16.97	0.92(6.6)	1.64	16.99	0.92(7.1)
29	13.26	1.49(1.1)	4.87	13.20	1.49(1.2)
30	14.45	1.04(6.3)		14.73	0.83(6.4)

^aData are for solutions in CDCl₃ + 1 drop MeOH-*d*₄.

^bData are for solutions in CDCl₃. Coupling constants (in Hertz) are in parentheses.

^cAcetates: δ 171.18 and 21.72 (2.00); 170.42 and 21.16 (2.01); 170.38 and 21.36 (2.07); 169.98 and 21.38 (2.10).

dard. Mass spectra were measured on a VG 70-25S mass spectrometer operating at 70 eV.

P. acerosa (dry wt 675 g) was collected in August 1987 at Lau's reef (-10 m), Tobago. The collection and identification were made by Mr. Richard Laydoo of the Institute of Marine Affairs, Trinidad and Tobago, where a voucher specimen has been kept. The sample was immediately stored in Me₂CO for transportation to the laboratory. In the laboratory it was blended with fresh Me₂CO (12 liters), and the solvent was evaporated to provide an aqueous suspension, which was thoroughly extracted with CHCl₃. The extract yielded a dark red viscous oil (91 g) which was chromatographed on Si gel, with hexane-Me₂CO (3:1) and CHCl₃-MeOH (95:5) as eluents, to give seven major fractions. The most polar seventh fraction was rechromatographed on Si gel with CHCl₃-MeOH (95:5) to give acerosterol [2] (50 mg).

ACEROSTEROL [2].—Mp 118–120°; [α]_D -3.3° (c=0.04, MeOH); ir 3440 cm⁻¹; eims *m/z* (% rel. int.) [M]⁺ 476 (5), 458 (8), 440 (6), 422 (5), 346 (13), 335 (10), 142 (64), 69 (100); hreims 476.3852 (calcd for C₃₀H₅₂O₄, 476.3866).

Acerosterol 2 (20 mg) was acetylated with a mixture of Ac₂O (0.5 ml) and pyridine (1 ml) for 6 h at 80°. The solvent was removed in a stream of N₂ to give the tetraacetate 3 (23 mg) as a colorless gum: [α]_D -22.0° (c=0.15, CHCl₃); ir 1726 cm⁻¹; eims *m/z* (rel. int.) [M]⁺ 644 (1), 584 (7), 524 (5), 481 (7), 464 (6), 404 (8), 361 (23), 69 (100); hreims 644.4347 (calcd for C₃₈H₆₀O₈, 644.4288).

ACKNOWLEDGMENTS

The Centre at the University of Guyana was funded by the Canadian International Develop-

ment Agency. Research at the University of Toronto was supported by grants from the Natural Sciences and Engineering Research Council of Canada.

LITERATURE CITED

1. W. Fenical, *J. Nat. Prod.*, **50**, 1001 (1987).
2. W.F. Tinto, W.R. Chan, W.F. Reynolds, and S. McLean, *Tetrahedron Lett.*, **31**, 464 (1990).
3. W.F. Tinto, L. John, A.J. Lough, W.F. Reynolds, and S. McLean, *Tetrahedron Lett.*, **32**, 4661 (1991).
4. W.R. Chan, W.F. Tinto, R.S. Laydoo, P.S. Manchand, W.F. Reynolds, and S. McLean, *J. Org. Chem.*, **56**, 1773 (1991).
5. W.F. Tinto, L.M.D. John, W.F. Reynolds, and S. McLean, *Tetrahedron*, **46**, 8679 (1991).
6. Y. Shimizu, M. Alam, and A. Kobayashi, *J. Am. Chem. Soc.*, **98**, 1059 (1976).
7. W.F. Reynolds, S. McLean, M. Perpich-Dumont, and R.G. Enriquez, *Magn. Reson. Chem.*, **27**, 162 (1989).
8. J. Finer, J. Clardy, A. Kobayashi, M. Alam, and Y. Shimizu, *J. Org. Chem.*, **43**, 1990 (1978).
9. A.Y.L., Shu and C. Djerassi, *Tetrahedron Lett.*, **23**, 4627 (1981).
10. N. Withers, in: "Marine Natural Products, Chemical and Biological Perspectives." Ed. by P.J. Scheuer, Academic Press, New York, 1983, Vol. 5, pp. 87–130.
11. N.W. Withers, W.C.M.C. Kokke, W. Fenical, and C. Djerassi, *Proc. Natl. Acad. Sci. U.S.A.*, **79**, 3764 (1982).

Received 17 June 1992